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# Geological Sequestration of CO<sub>2</sub> A Brief Overview and Potential for Application for Oklahoma

Presented to Oklahoma Clean Lakes and Watersheds Association Quartz Mountain Resort April 1, 2009

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#### **Outline of Presentation**

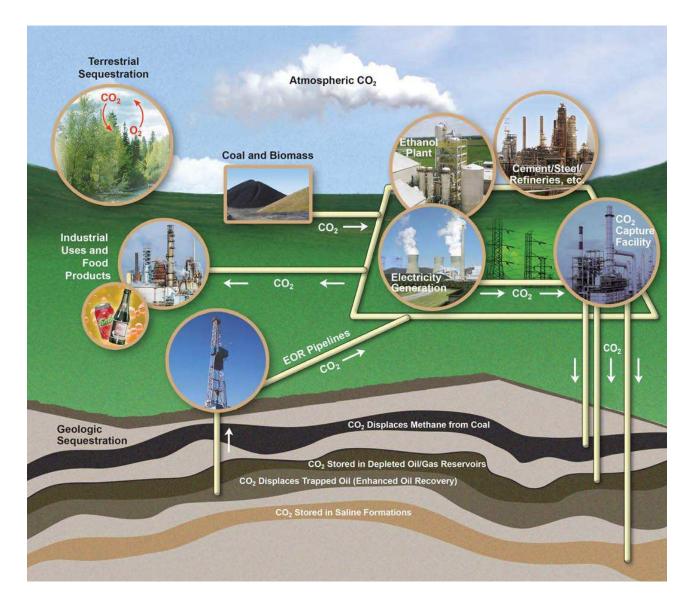
- Brief Overview of Geological Sequestration
- Potential for CO2-EOR in Oklahoma
- Research at R.S. Kerr Research Center, Ada, OK

#### Disclaimer:

This presentation does not necessarily reflect EPA policy.



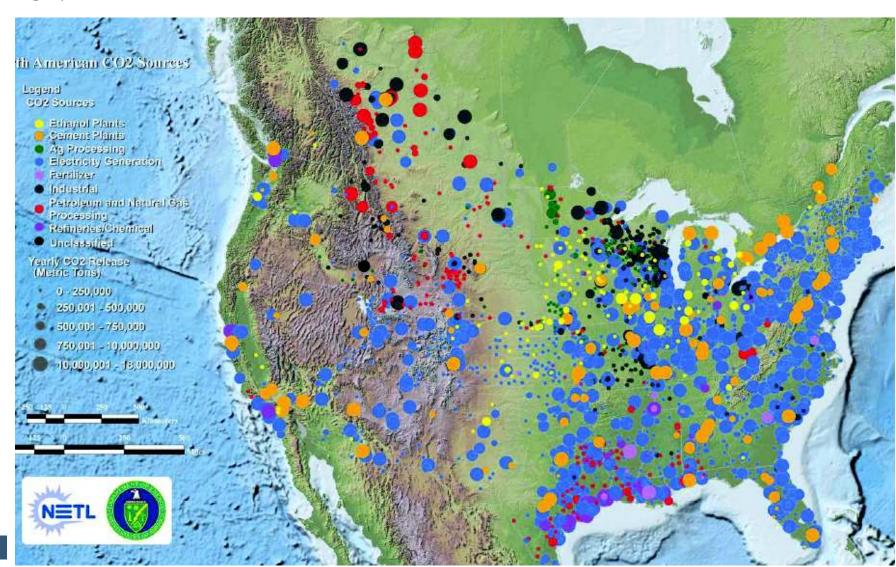
Carbon Capture and Storage (CCS) and Geological Sequestration (GS)



From DOE, 2008

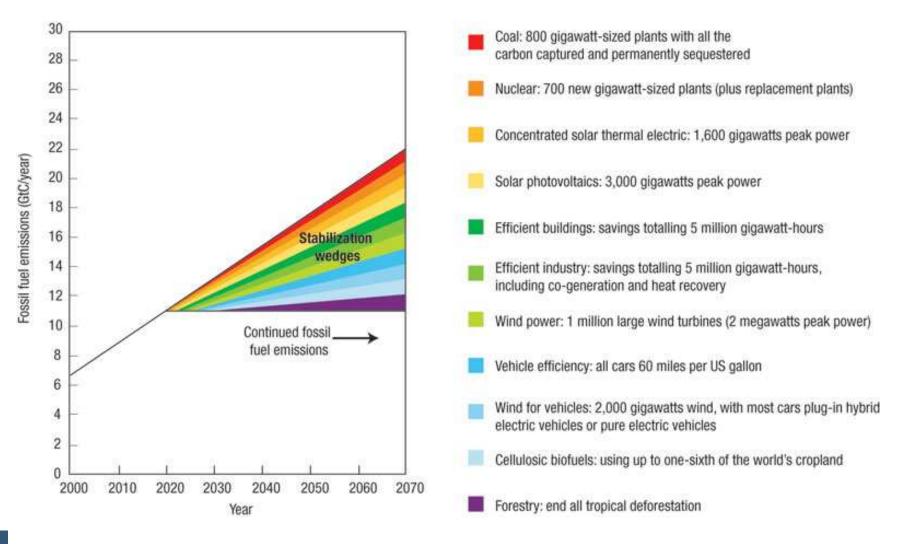


## **Stationary Sources of CO<sub>2</sub>**





## CCS as Part of the Big Picture

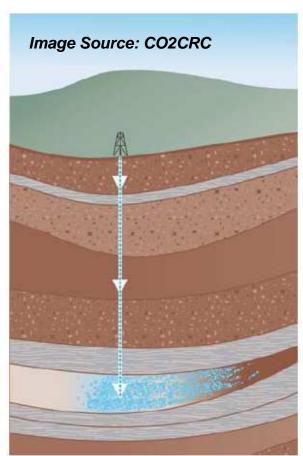


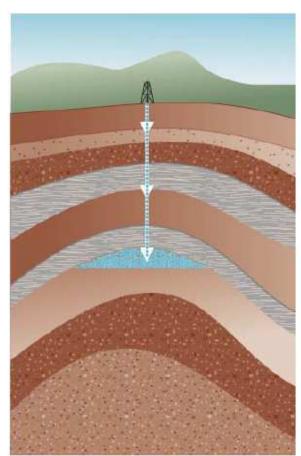


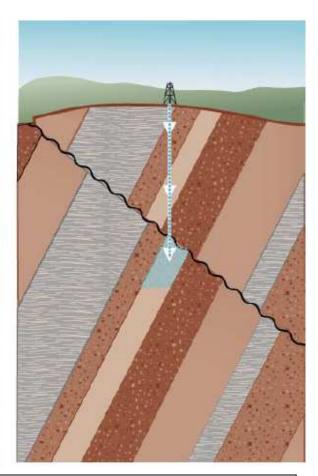
## **Trapping Mechanisms**

The buoyancy of supercritical CO<sub>2</sub> necessitates that the injection zone must be overlain by a primary confining system of sufficient regional thickness and lateral extent to contain the entire CO<sub>2</sub> plume and associated pressure front. This will initially be the primary mechanism of containment.

# Stratigraphic and Structural Trapping





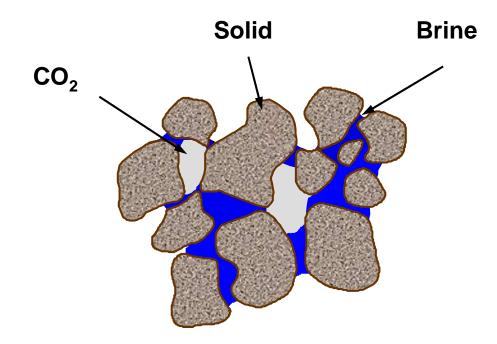


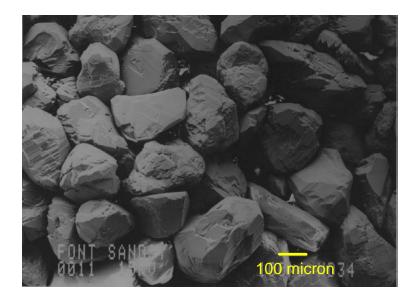
In stratigraphic trapping (left), CO<sub>2</sub> is trapped by an overlying layer of cap rock coupled with impermeable rock within a narrowing of the storage formation.

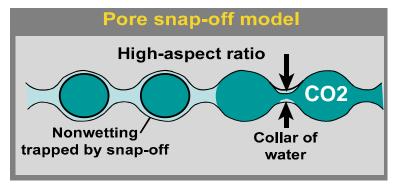
In structural trapping, CO<sub>2</sub> is trapped by a fold in the rock formations (middle) or by impermeable rock layers shifted along a sealing fault (right) to contain the CO<sub>2</sub>.



## **Capillary Trapping**





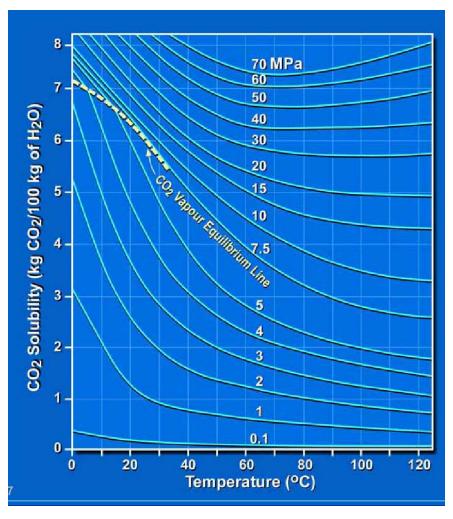


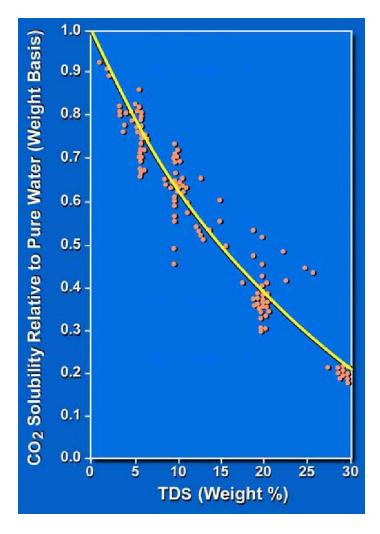
From Bryant, 2005



## Dissolution/Ionic Trapping

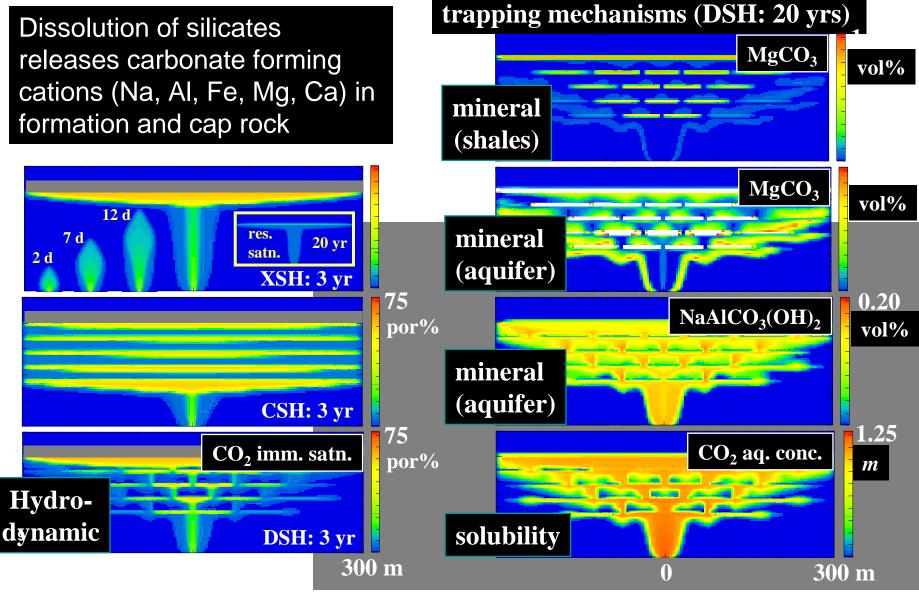
$$CO_2(g) + H_2O \leftrightarrow H_2CO_3 \leftrightarrow HCO_3^- + H^+ \leftrightarrow CO_3^{2-} + 2H^+$$





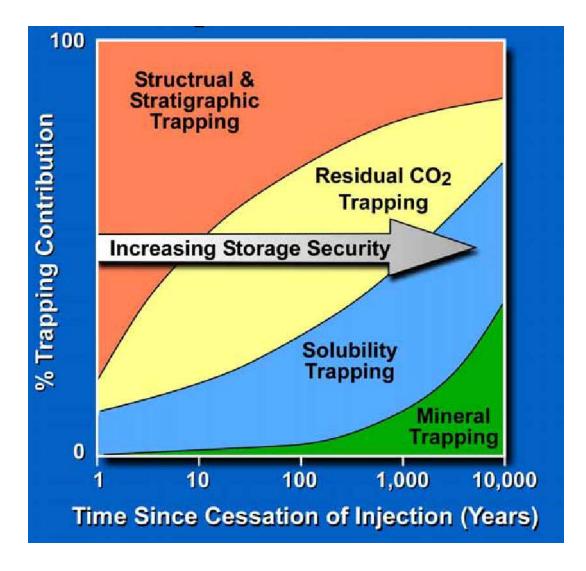


releases carbonate forming cations (Na, Al, Fe, Mg, Ca) in formation and cap rock





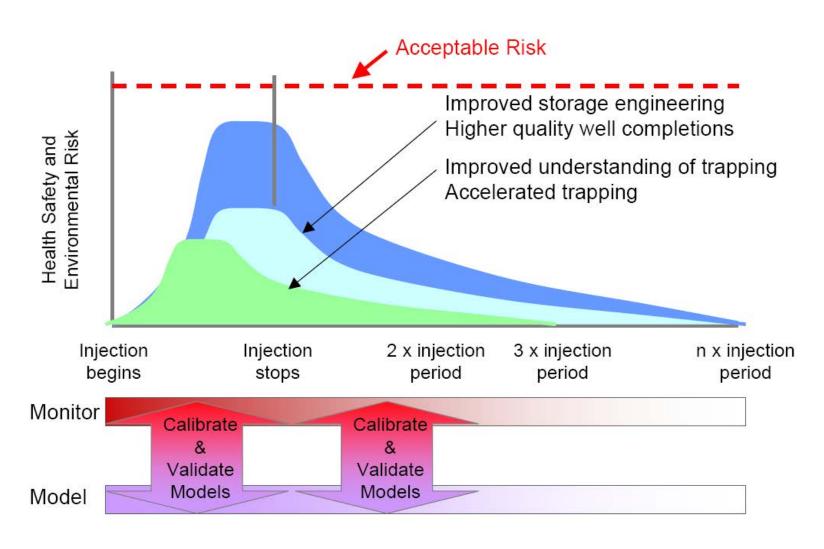
Primary
Mechanisms of
CO<sub>2</sub> Storage (not
including
adsorption to coal
and shale)



From IPCC Special Report on Carbon dioxide Capture and Storage



## Improved Design = Less Liability





## **Storage Formations**



# Deep Saline Aquifers





# "Unmineable Coal Seams

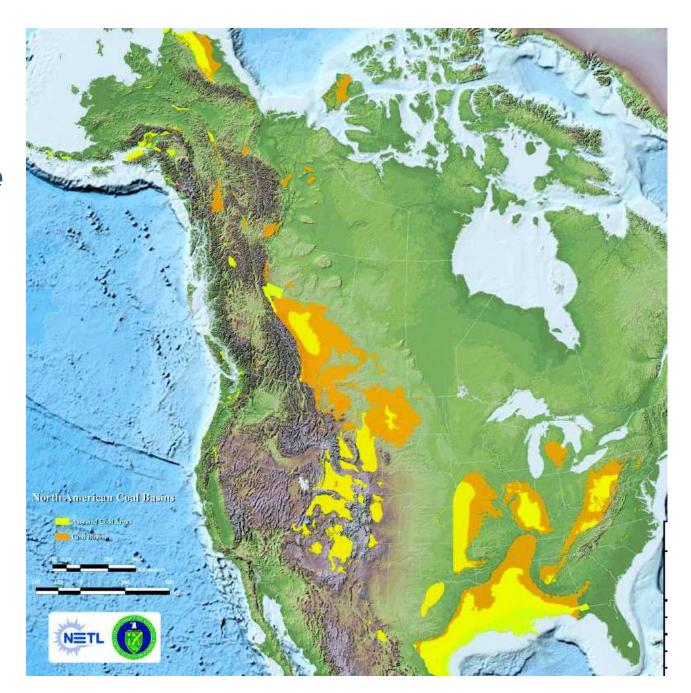


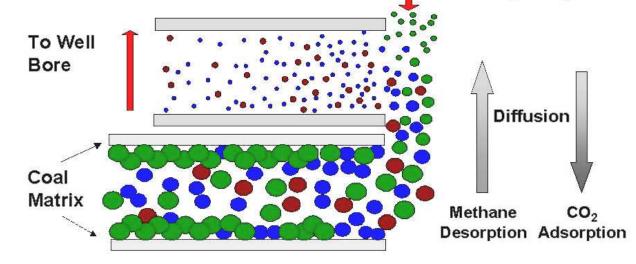
Figure from DOE, 2007



- · CO2 displaces methane after injection
- No breakthrough of CO<sub>2</sub> until saturation
- · Higher cleat pressure results in faster flow

CO<sub>2</sub> Water
Injection Methane
CO<sub>2</sub>

## Enhanced Coal Bed Methane (ECBM) Recovery



- Value added incentive.
- Access to "unmineable" coal deposits.
- Potential 1.5% storage capacity in U.S.

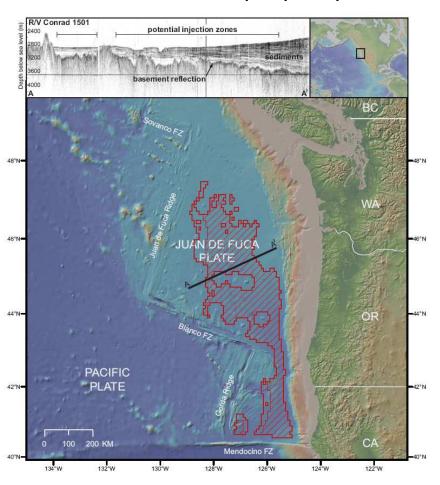
Figure from DOE, 2006

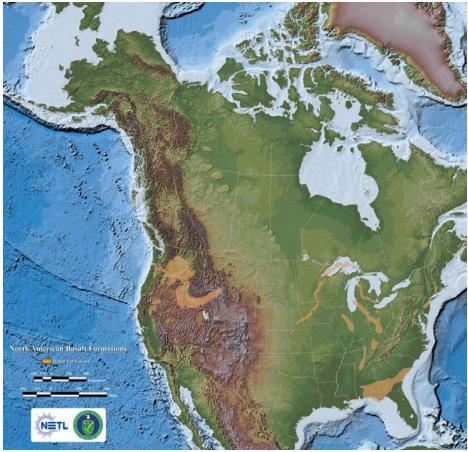
- Swelling reduction of permeability.
- Brine disposal.
- Close to or part of USDWs
- Many coal seams too shallow for storage as supercritical fluid.



## Potential Storage in Basalt

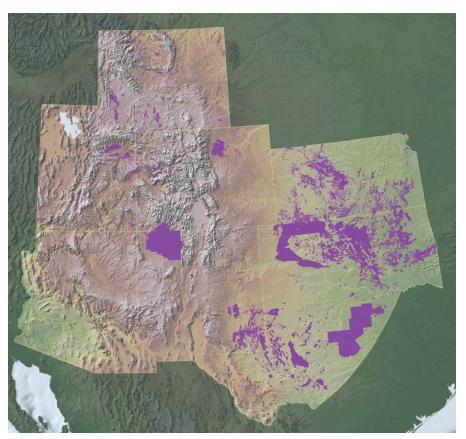
The release of Ca<sup>2+</sup> and Mg<sup>2+</sup> ions from silicate minerals in basalt could result in rapid precipitation of carbonate minerals.

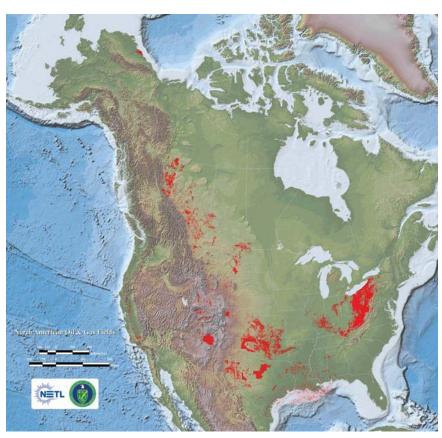






## Storage Capacity in Petroleum and Gas Reservoirs





Most of the storage capacity in Oklahoma associated with oil and gas reservoirs.

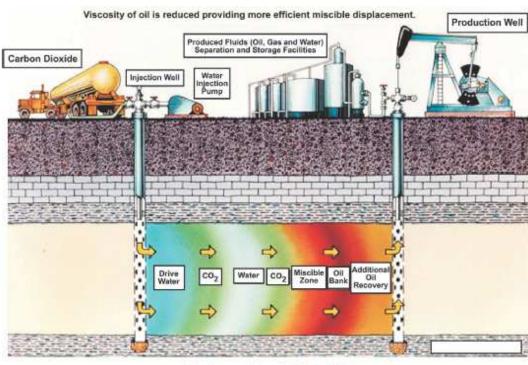


## Enhanced Oil Recovery (EOR)

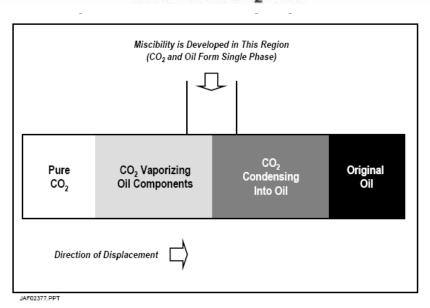
Figure from DOE, 2007

During miscible displacement, CO<sub>2</sub> lowers the viscosity and interfacial tension of oil. Deep "light" oil.

Figures from Basin Oriented Strategies for CO<sub>2</sub> Enhanced Oil Recovery: Oklahoma, DOE, NETL, March 2005



Schematic of CO<sub>2</sub> EOR



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## CO<sub>2</sub>-EOR



## Benefits of CO<sub>2</sub>-EOR

- Value Added Incentive. Revenue for petroleum corporations from additional production of petroleum and carbon credits. Revenue for corporations or utilities selling CO<sub>2</sub> to meet cap and trade requirements.
- Tax Revenue. Substantial increase in tax revenue for States.
- **Trade Imbalance**. Reduce trade imbalance for imported petroleum with decreased reliance on countries hostile to U.S.
- **Infrastructure**. Some existing infrastructure already present. Widespread application will create additional infrastructure such as pipelines for eventual injection into saline aquifers.
- Mineral Rights. Mineral rights at petroleum reservoirs are already established.
- **Site Characterization**. Unlike saline aquifers, petroleum reservoirs are already well characterized.
- Carbon Neutrality. CO<sub>2</sub>-EOR is 70% carbon neutral. Improve design to achieve 100% carbon neutrality (Corn-based ethanol only 10-15% carbon neutral and a net contributor to CO<sub>2</sub> emissions when coal used as a process fuel).

Taken from: Storing CO2 with Enhanced Oil Recovery, DOE/NETL-402/1312/02-07-08



Nationwide CO2-EOR application would add about 85 billion barrels of incremental domestic oil supply ~ (4X current proved reserves).

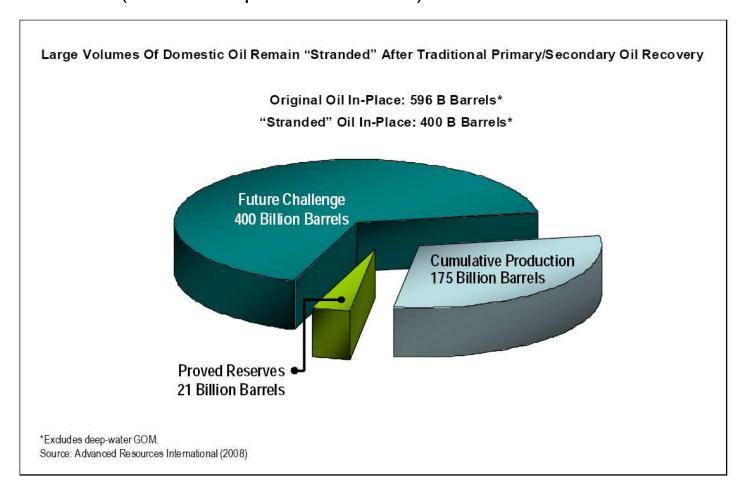


Figure from: Storing CO2 with Enhanced Oil Recovery, DOE/NETL-402/1312/02-07-08



As of 2006, there were 86 CO2-EOR project in the U.S.

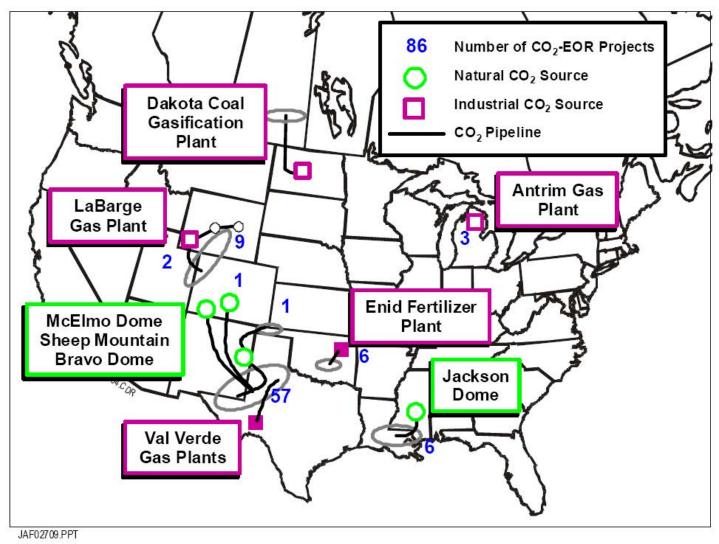


Figure from: Storing CO2 with Enhanced Oil Recovery, DOE/NETL-402/1312/02-07-08





CO<sub>2</sub> could be provided by H<sub>2</sub> plants in Ponca City (11 MMcfd, Ardmore (26 MMcfd, and Wynnewood (9 MMcfd)

# Conceptual Pipeline System Connecting Fossil-Fuel Based CO<sub>2</sub> Sources with Major Petroleum Reservoirs

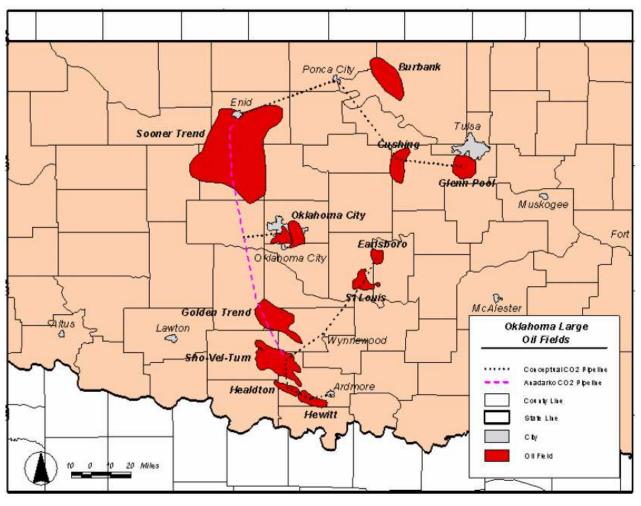


Figure from Basin Oriented Strategies for CO<sub>2</sub> Enhanced Oil Recovery: Oklahoma, DOE, March 2005



# History of Oklahoma Crude Oil Production, 1950-2002



Figure from Basin Oriented Strategies for CO<sub>2</sub> Enhanced Oil Recovery: Oklahoma, March 2005



Figure 1. Impact of Advanced Technology and Improved Financial Conditions on Economical Recoverable Oil from Oklahoma's Major Reservoirs Using CO<sub>2</sub>-EOR (Million Barrels).

Technical recovery at 63 large reservoirs (60.5% of state) is about 5.4 billon barrels.

Technical recovery in entire state is about 9 billion barrels from 45 billion barrels remaining. OOIP was 60 billion barrels.

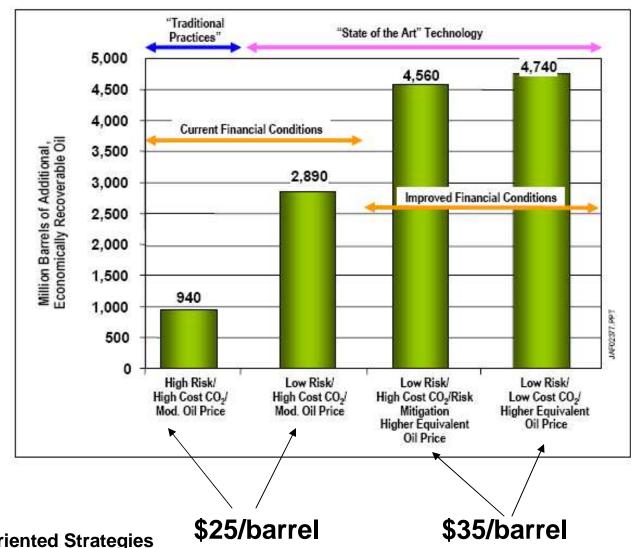


Figure from Basin Oriented Strategies for CO2 Enhanced Oil Recovery: Oklahoma, March 2005





Friday, July 25, 2008

## Federal Rule-Making

120 day comment period

#### Part II

## **Environmental Protection Agency**

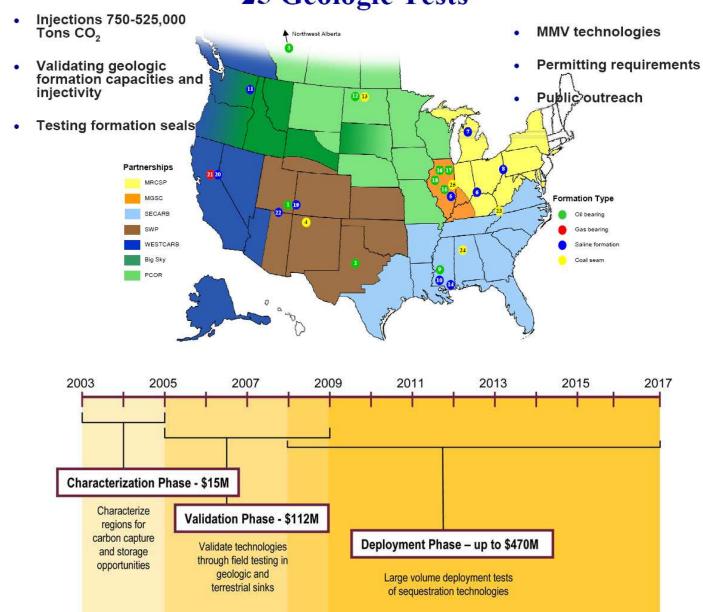
40 CFR Parts 144 and 146 Federal Requirements Under the Underground Injection Control (UIC) Program for Carbon Dioxide (CO<sub>2</sub>) Geologic Sequestration (GS) Wells; Proposed Rule



## **DOE Research**



## Phase II Field Validation 25 Geologic Tests





# Summary of Potential Monitoring Tools

Onshore only Offshore only Onshore & Offshore Secondary use Primary use 3D/4D surface seismic Time lapse 2D surface seismic Multicomponent seismic Acoustic Boomer / Sparker Seismic imaging High resolution acoustic imaging Microseismic monitoring 4D cross-hole seismic Well based 4D VSP Sidescan sonar Sonar Bathymetry Multi beam echo sounding Time lapse surface gravimetry Gravimetry Time lapse well gravimetry Surface EM Seabottom EM Electric / Electro -Cross-hole EM magnetic Permanent borehole EM Cross-hole ERT ESP Downhole fluid chemistry Fluids PH measurements Tracers Geochemical Marine Seawater chemistry **Bubble stream chemistry** Short closed path (NDIRs & IR Short open path (IR diode lasers) Atmos-phere Long open path (IR diode lasers) **Eddy covariance** Gas flux Soil Gas concentrations **Ecosystems Ecosystems studies** Airborne hyperspectral imaging Remote sensing Satellite interferometry Airborne EM Geophysical logs Others Pressure / temperature **Tiltmeters** 

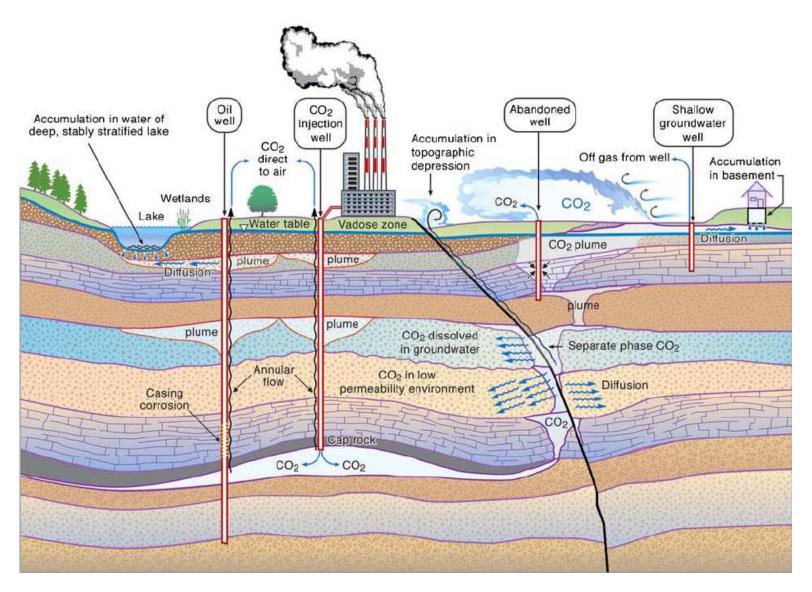


## Research at R.S. Kerr Research Center

## Potential Leakage Pathways and Consequences

United States
Environmental Protection
Agency

From Benson and Hepple (2005)



#### Use Soil-Gas, Gas Flux, and Ground-Water Monitoring to Evaluate the Potential Leakage from Well Penetrations (NRMRL-Ada)

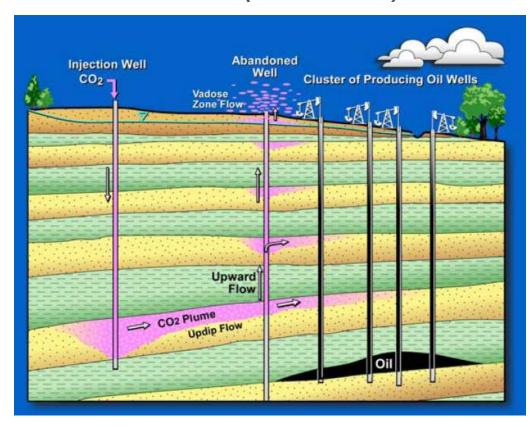


Figure from Celia et al. (2006)

Use soil-gas, gas flux, and ground-water monitoring of C1-C4 hydrocarbons,  $CO_2$ ,  $\delta$ 13C,  $\Delta$ 14C,  $H_2$ , He,  $H_2$ S,  $^{222}$ Rn, major ions, pH, and inorganics to evaluate the presence of existing migration pathways.

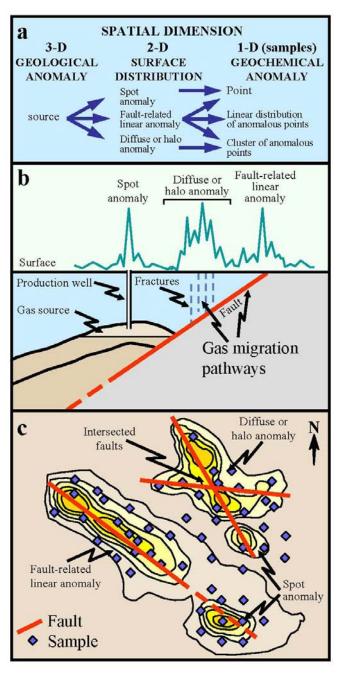


Figure modified from Ciotoli et al., (2004).



### Evaluate Impacts to USDWs due to Carbon Dioxide Release from Geologic Sequestration Projects: Modeling and Experimental Studies (NRMRL-Ada)

- Conduct column and batch-scale studies from formation (USDW) samples collected from test sites.
- Examine and simulate element partitioning and associated kinetics between the solid and aqueous phase over a range of CO<sub>2</sub> partial pressures.
- Where appropriate, modify geochemical databases with the most current thermochemical data.
- Use results to prepare sampling strategies for a controlled CO<sub>2</sub> injection field study.





Use semi-analytical solutions to create a user-friendly, open-source software package to allow a rapid, inexpensive, first-order determination of the Area of Review

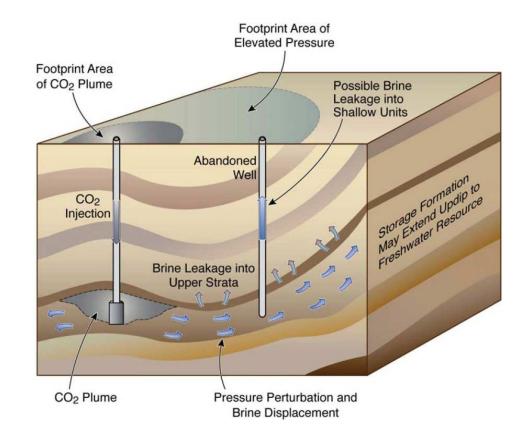
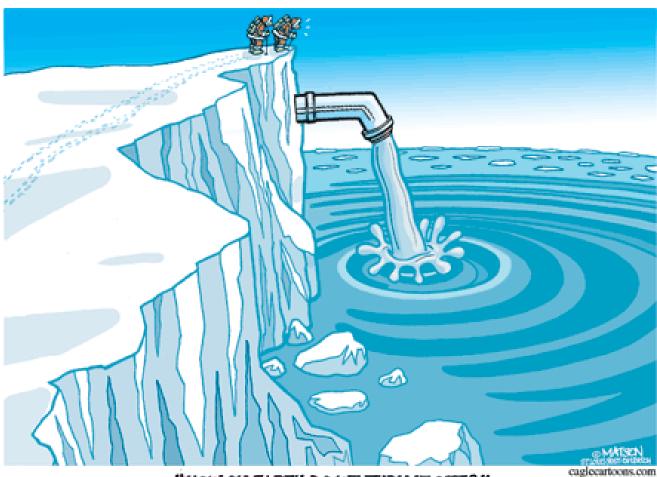


Figure from Birkholzer et al., 2008

- Solutions will complement not replace need for numerical analysis.
- Can be used for designing monitoring strategies (e.g., pressure perturbation in overlying permeable formation).
- Computationally more efficient for evaluating leakage through abandoned wells



## **Questions?**



"HOW ON EARTH DO WE TURN IT OFF?"